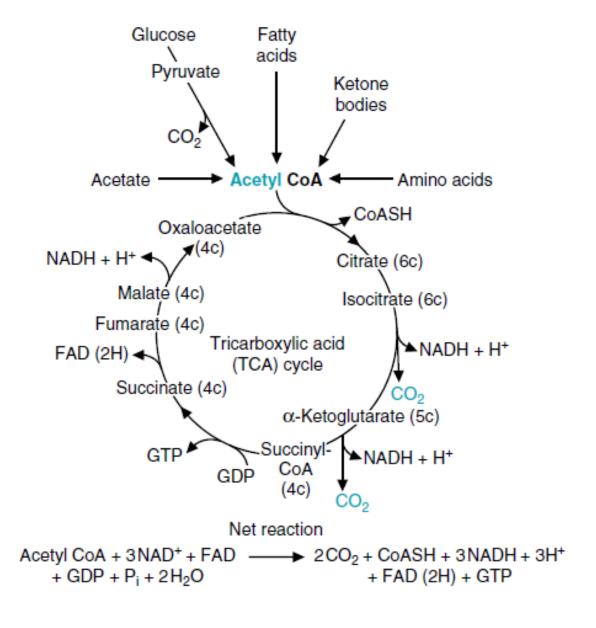


The Central Metabolic Hub

Prof. Nafez Abu Tarboush

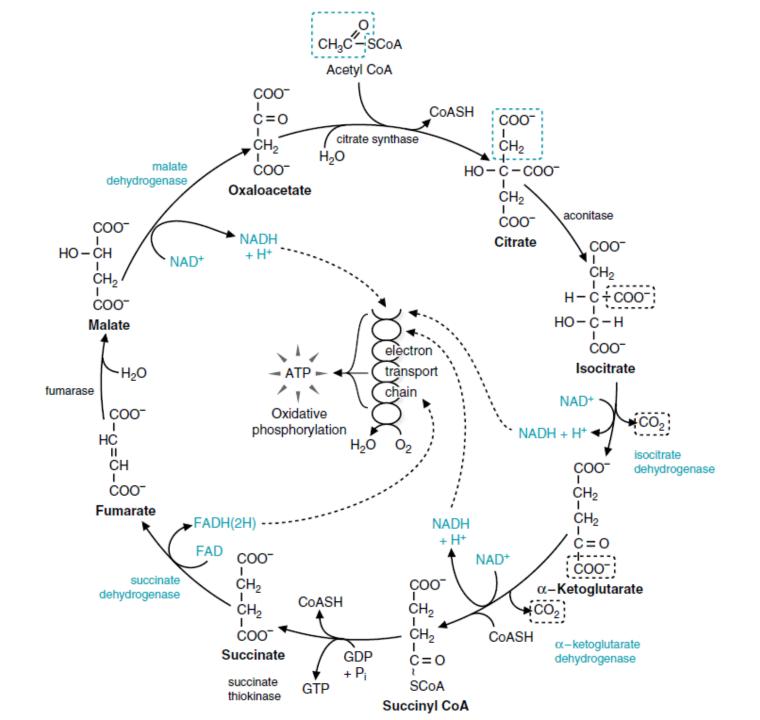


COMPONENTS & STEPWISE REACTIONS

CIA Sent Soldiers For My Office



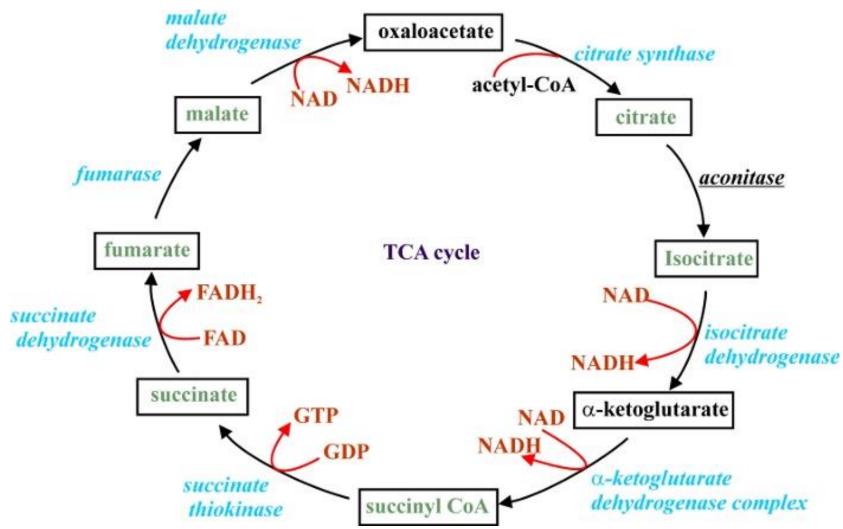
COMPONEN





ENZYMES OF THE TCA CYCLE

- Citrate synthase
- Aconitase
- Isocitrate DH
- α -ketoglutarate DH
- Succinate thio-kinase
- Succinate DH
- Fumarase
- Malate DH





FORMATION OF CITRATE

What drives the reaction forward?

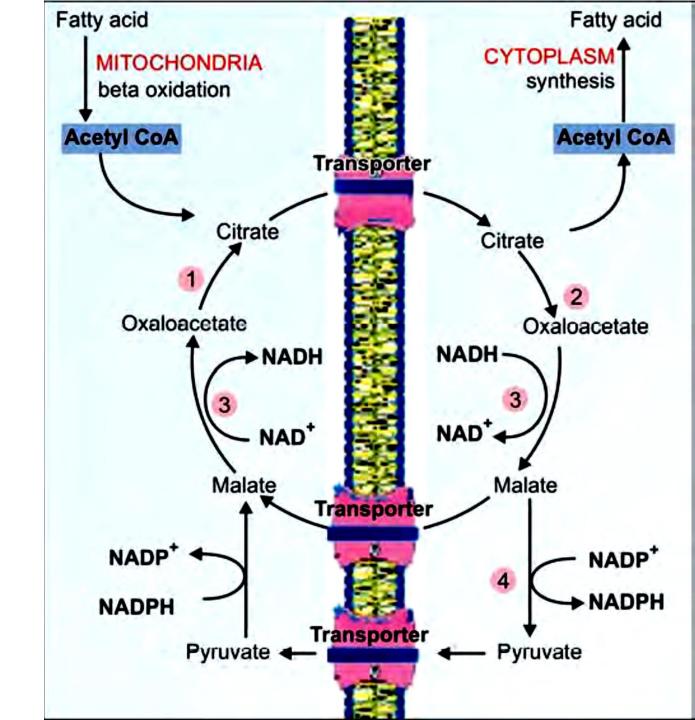
Is it reversible or irreversible?

Can it be reversed?

ATP-Citrate lyase or ATP-Citratase!

Activated by insulin and is highly expressed in lipogenic tissues like the liver and adipose tissue

Link to cancer metabolism



FORMATION AND OXIDATION OF ISOCITRATE

CH,-COO

CH2-COO

Cis-Aconitate

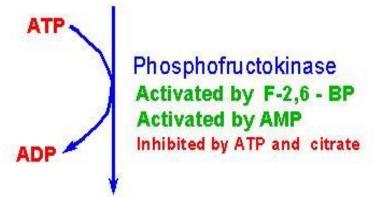
2 Aconitase

H₂O 2 Aconitase

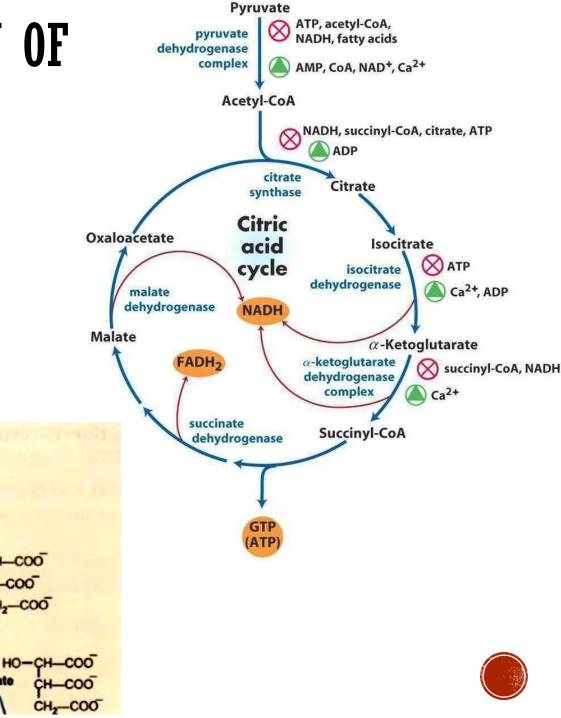
Citric acid

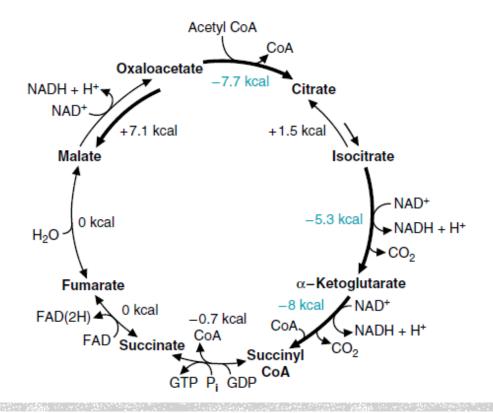
Control at the committed step of glycolysis

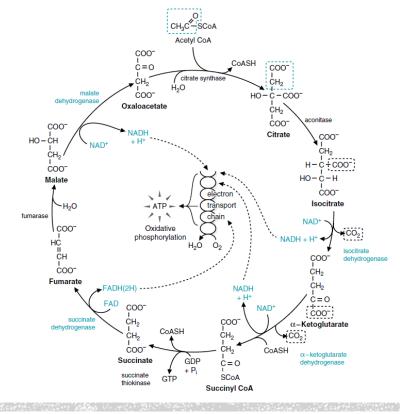
Fructose 6 - phosphate



Fructose 1,6 - bisphosphate







α-KETOGLUTARATE TO SUCCINYL COA

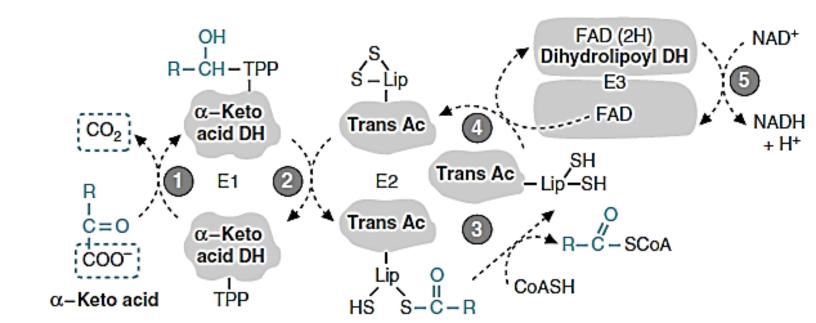
- Oxidative decarboxylation
- Thiamine pyrophosphate, lipoic acid, and FAD
- Keto group oxidized to acid, CoA-SH, succinyl CoA
- Energy conserved as NADH, thioester bond
- The highest energy vield provided!



α-KETOACID DEHYDROGENASE COMPLEXES (TLCFN)

 $(\alpha$ -ketoglutarate, pyruvate, and branched chain α keto acid) dehydrogenase complexes Huge enzyme complexes, multiple subunits of 3 different enzymes (no loss of energy, substrates for E2 and E3 remain bound → higher rate)

■ E1, E2, & E3 are a decarboxylase (TPP), a transacylase (lipoate), & a dehydrogenase (FAD)



δ COO-

α C=0

α-Ketoglutarate

δ COO

γ CH₂ β CH₂

Succinyl CoA

α C-SCoA

Thiamine—PP Lipoate

NAD+

NADH

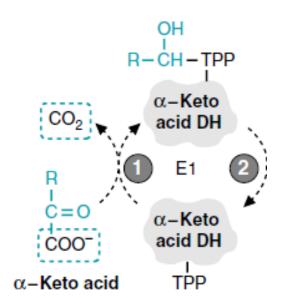
+ H+

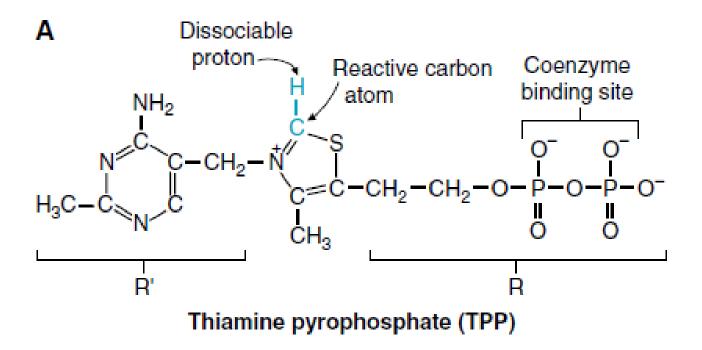
CoASH-

γ CH₂

THIAMINE PYROPHOSPHATE

Thiamine deficiency, α -ketoglutarate, pyruvate, & branched chain α -keto acids accumulate in the blood

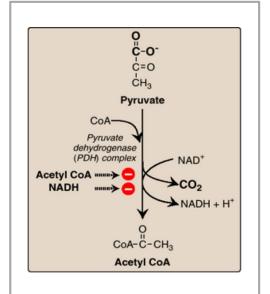


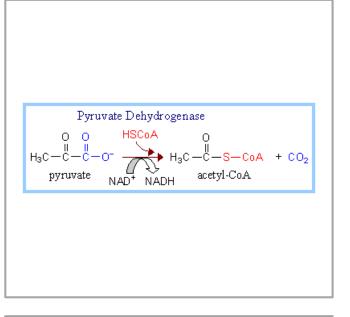


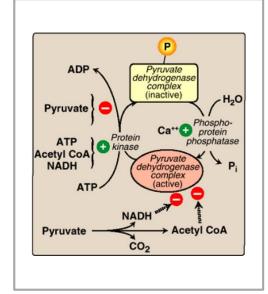


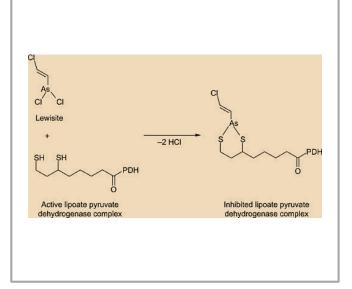
OXIDATIVE DECARBOXYLATION OF PYRUVATE

- Component enzymes
- Coenzymes
- Regulation of the pyruvate dehydrogenase complex
 - Pyruvate dehydrogenase deficiency: A deficiency in E₁ component is the most common biochemical cause of congenital lactic acidosis (X-linked, no treatment)
- Mechanism of arsenic poisoning







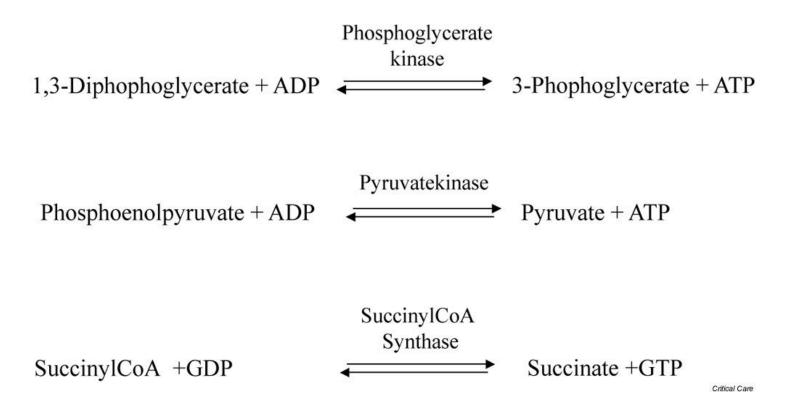


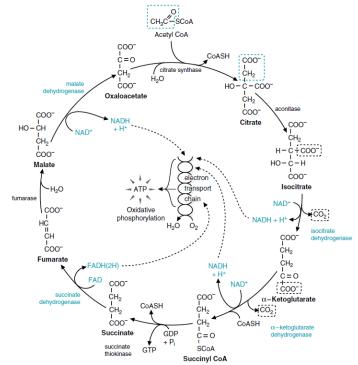


GENERATION OF GTP

-Succinyl CoA thioester bond, succinate thiokinase, substrate level phosphorylation

$GTP + ADP \leftrightarrow GDP + ATP$

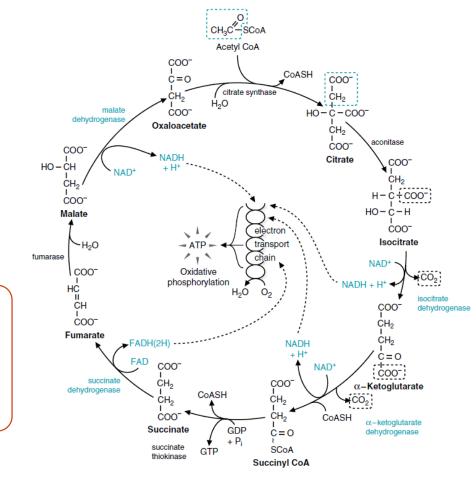




OXIDATION OF SUCCINATE TO OXALOACETATE

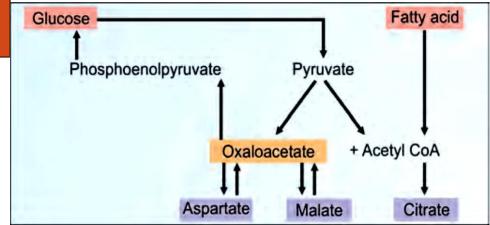
Oxidation of succinate to fumarate, succinate dehydrogenase, FAD

Fumarase, OH + H⁺ from water, fumarate to malate Alcohol group of malate oxidized to a keto group, NADH



Oxaloacetate as a Junction Point

- Viewed as a catalyst
- An important junction point in metabolism

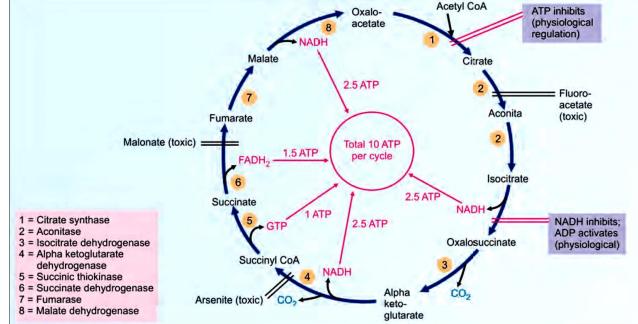


BIOENERGETICS OF TCA CYCLE

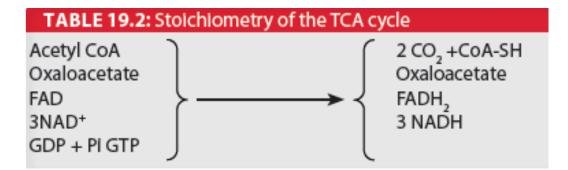
- Like all pathways, overall net $-\Delta G$ (-228 kcal/mole), not 100%
- NADH, FAD(H2), and GTP (10ATP), 207 Kcal, 90%
- Three reactions have large (-ve) values
- Physiologically irreversible, low products

_	Oxaloacetate Oxaloacetate Oxaloacetate	
kcal/mole	NAD+ +7.1 kcal +1.5 kcal	
3 NADH: 3 × 53 = 159 1 FAD(2H) = 41 1 GTP = 7 Sum = 207	Malate Isocitrate NAD+ NAD+ NADH + H+ CO ₂ Fumarate FAD(2H) O kcal FAD Succinate O kcal O kcal	

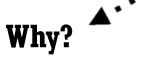
	TABL	TABLE 19.1: ATP generation steps				
	Step No	Reactions	Co-enzyme	ATPs (old- calculation)	ATPs (new calculation)	
	3	Isocitrate → alpha keto glutarate	NADH	3	2.5	
	4	Alpha keto glutarate → succinyl CoA	NADH	3	2.5	
	5	Succinyl CoA→Succinate	GTP	1	1	
	6	Succinate → Fumarate	FADH ₂	2	1.5	
	8	Malate → Oxalo acetate	NADH	3	2.5	
			Total	12	10	



NET RESULT OF THE CYCLE & ITS SIGNIFICANCE







Box 19.1: Significance of citric acid cycle

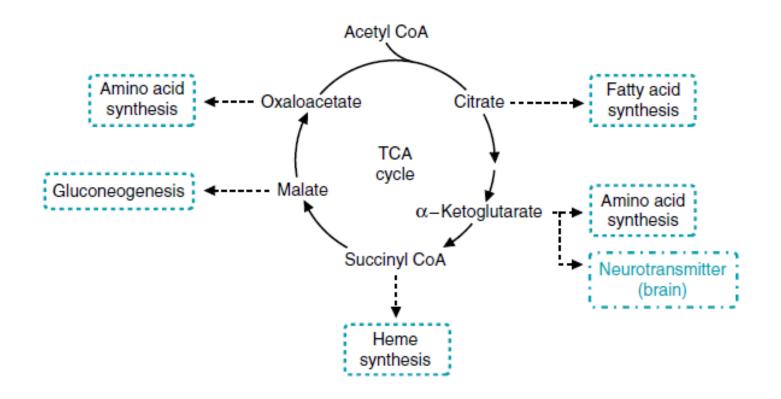
- Complete oxidation of acetyl CoA
- ATP generation
- Final common oxidative pathway
- Integration of major metabolic pathways
- 5. Fat is burned on the wick of carbohydrates
- 6. Excess carbohydrates are converted as neutral fat
- No net synthesis of carbohydrates from fat
- Carbon skeletons of amino acids finally enter the citric acid cycle
- 9. Amphibolic pathway
- Anaplerotic role.
- Fats are burned in the fire of carbohydrates
- ✓ Fat cannot be converted to glucose because pyruvate dehydrogenase reaction is an irreversible step





FAT IS BURNED ON THE WICK OF CARBOHYDRATES

- What Happens in a Low-Carbohydrate State? (The "Fire Goes Out")
 - OAA is Drained and Pyruvate is Diverted
- Krebs Cycle Stall!!!
- The Pathological Consequences: Ketogenesis
- Clinical Correlations
 - Diabetic Ketoacidosis (DKA), Starvation, Weight Loss Diets

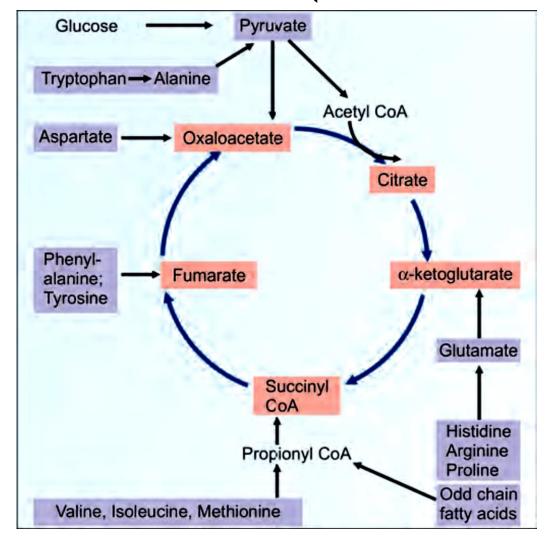


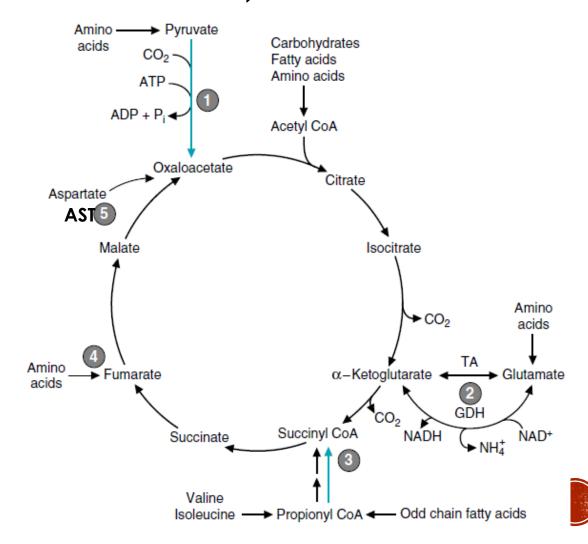
CATAPLEROSIS

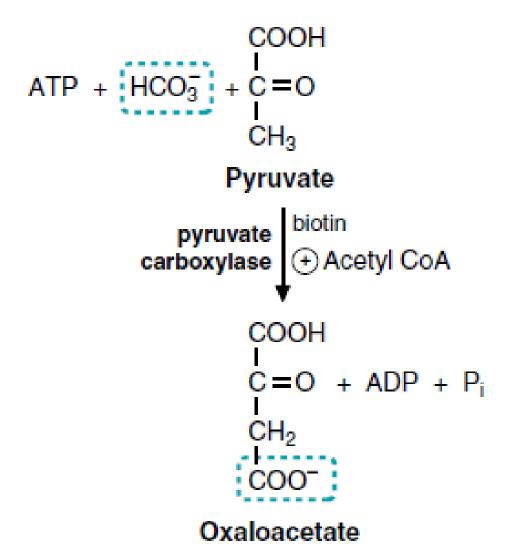
- Intermediates are Precursors for Biosynthetic Pathways
 - (citrate, acetyl CoA, fatty acid synthesis, liver)
 - (fasting, malate, gluconeogenesis, liver)
 - (Succinyl CoA, heme biosynthesis, bone marrow)
 - (α -ketoglutarate, glutamate, GABA, a neurotransmitter, brain)
 - (α -ketoglutarate, glutamine, skeletal muscle to other tissues for protein synthesis)
 - (OAA, Aspartate, Asparagine, Gluconeogenesis, liver)



ANAPLEROTIC ROUTES (AMINO ACID DEGRADATION)



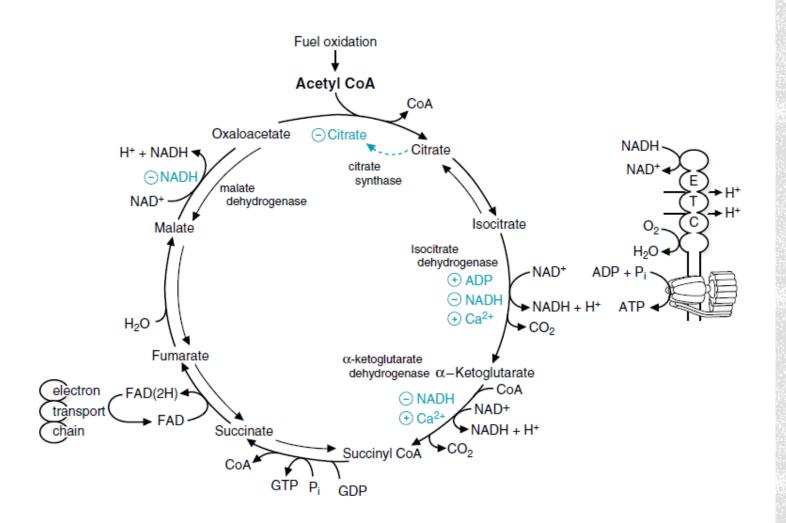




ANAPLEROTIC REACTIONS

- Pathways or reactions that replenish the intermediates of the TCA cycle
- Pyruvate Carboxylase is a major anaplerotic enzyme (requires biotin)
- Found in many tissues, liver, kidneys, brain, adipocytes, and fibroblasts
- Very high conc. In liver and kidney (gluconeogenic pathway)
- Activated (acetyl CoA)



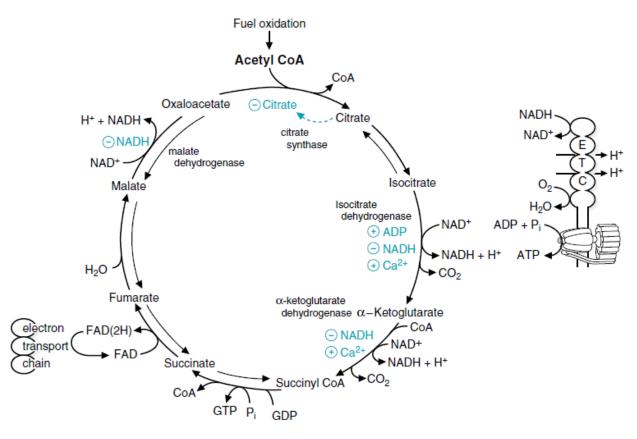


REGULATION OF THE TCA CYCLE

- Correspond to ETC (ATP/ADP)
- Two major messengers (feedback): (a) phosphorylation state of adenines, (b) the reduction state of NAD
- Adenine nucleotides pool and NAD pool are relatively constant



REGULATION — CITRATE & CITRATE SYNTHASE



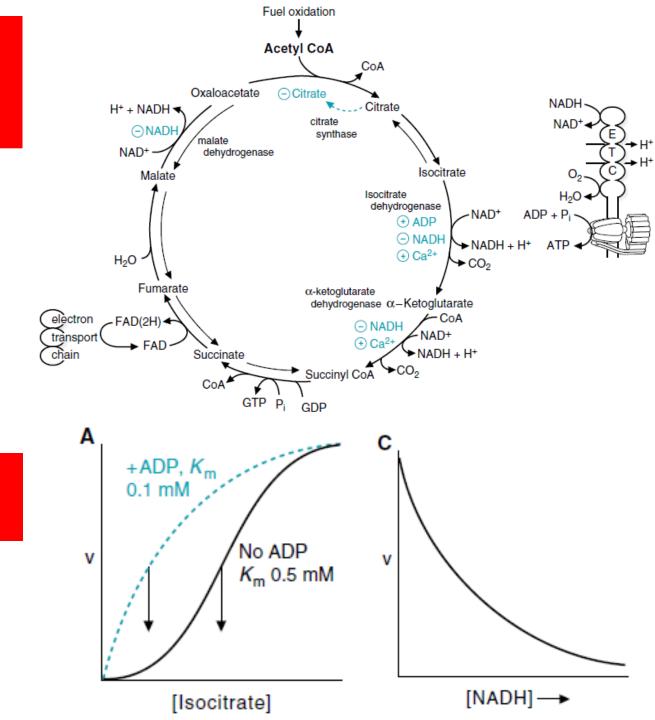
- Rate regulated by oxaloacetate & citrate (inhibitor)
- ATP: allosteric inhibitor
- Effect of citrate:
 - Allosterically inhibits PFK, the key enzyme of glycolysis
 - Stimulates fructose-1,6bisphosphatase, a key enzyme of gluconeogenesis
 - Activates acetyl CoA carboxylase, a key enzyme of fatty acid synthesis

ISOCITRATE DH

- Best regulation (rate-limiting)
- Allosterically: activated (ADP, Ca⁺²)
- Inhibition (NADH)
- No ADP vs. ADP (K_M) , a small change in ADP, great effect

α -Ketoglutarate DH

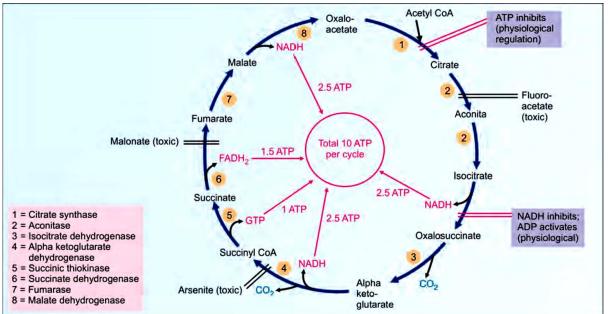
- Inhibited: NADH, succinyl CoA, GTP
- Activated: Ca⁺²



INHIBITORS OF TCA CYCLE (PHYSIOLOGICAL?)

 A. Aconitase (citrate to aconitate) is inhibited by fluoroacetate (noncompetitive inhibition)

- B. Alpha ketoglutarate dehydrogenase (alpha ketoglutarate to succinyl CoA) is inhibited by Arsenite (noncompetitive inhibition)
- C. Succinate dehydrogenase (succinate to fumarate) is inhibited by malonate (competitive inhibition)



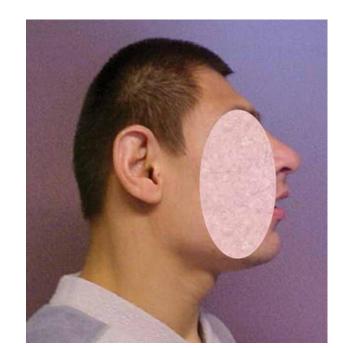


INBORN ERRORS OF WETABOLISM

• PDC Deficiency:

- Most common enzymatic cause of primary lactic acidosis
- Often presenting in neonates with severe neurological impairment
- Treatment involves a ketogenic diet to provide an alternative fuel source for the brain.

• Fumarase Deficiency: Causes severe encephalopathy, seizures, and developmental delay due to the accumulation of fumarate.





ONCOMETABOLITES AND CANCER:

• IDH Mutations:

- Common in gliomas and acute myeloid leukemia (AML)
- Confer a neomorphic activity that produces 2-hydroxyglutarate (2-HG): a competitive inhibitor of α-KGdependent dioxygenases, leading to a hypermethylation of DNA and histones
- This epigenetic dysregulation blocks cellular differentiation and promotes tumorigenesis.

SDH and FH Mutations:

- Cause hereditary paragangliomas and leiomyomatosis, respectively.
- Accumulating succinate or fumarate inhibits prolyl hydroxylases (PHDs), leading to the stabilization of HIF-1α under normal oxygen conditions (pseudohypoxia).
- This activates angiogenic and glycolytic programs, driving the Warburg effect and tumor growth.



NEURODEGENERATION

- Impaired mitochondrial function is the hallmark of neurodegenerative diseases like Parkinson's and Alzheimer's.
- The resulting energy deficit, oxidative stress, and disrupted calcium buffering contribute to excitotoxicity and neuronal death.

