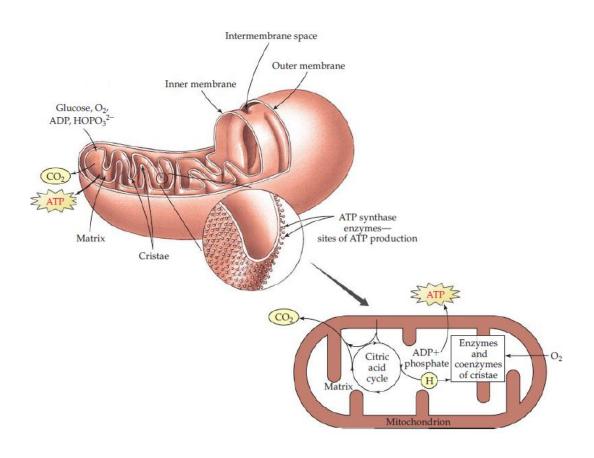


Prof. Nafez Abu Tarboush



MITOCHONDRIAL ARCHITECTURE

- OMM: <u>permeable</u> (MW<5,000); (transmembrane channels)
- IMM: <u>impermeable</u> (H⁺); specific transporters
- IMM: components of respiratory chain and ATP synthase
- Matrix: pyruvate dehydrogenase complex; TCA cycle enzymes, fatty acid β-oxidation pathway, and the pathways of amino acid oxidation
- Matrix contains all pathways of fuel oxidation except glycolysis

TABLE 20.3: Location of enzymes in mitochondria

Mitochondria, outer membrane:

Monoamino oxidase

Acyl CoA synthetase

Phospholipase A2

In between outer and inner membrane:

Adenylate kinase

Creatine kinase

Inner membrane, outer surface:

Glycerol-3-phosphate dehydrogenase

Inner membrane, inner surface:

Succinate dehydrogenase

Enzymes of respiratory chain

Soluble matrix:

Enzymes of citric acid cycle

Enzymes of beta oxidation of fatty acid

MITOCHONDRIAL MEMBRANES

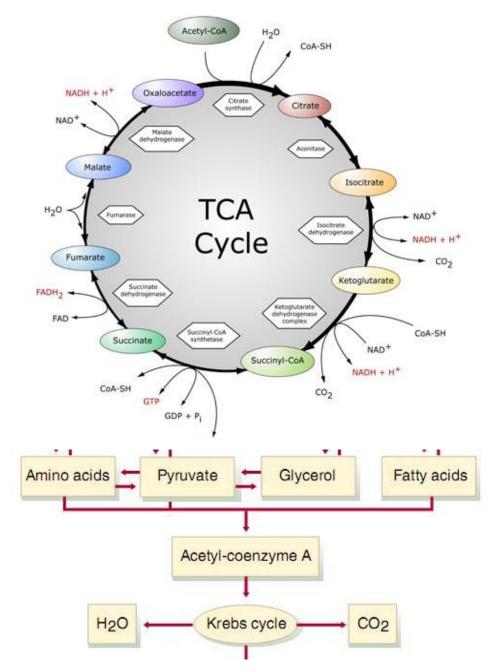
Inner membrane:

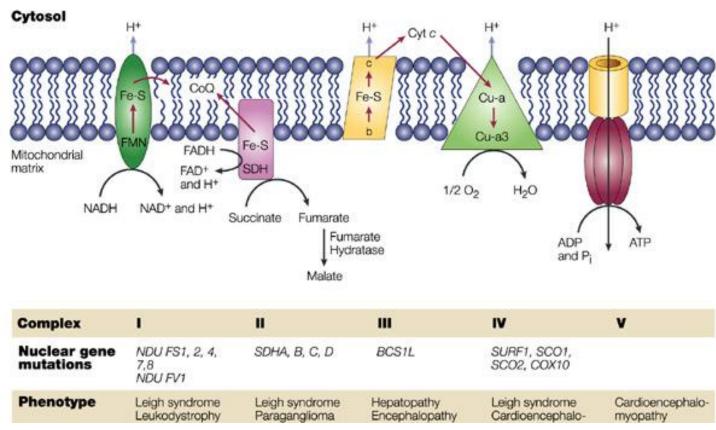
- 22% cardiolipin
- No cholesterol

• Outer membrane:

- Similar to cell membrane
- Less than 3% cardiolipin
- 45% cholesterol

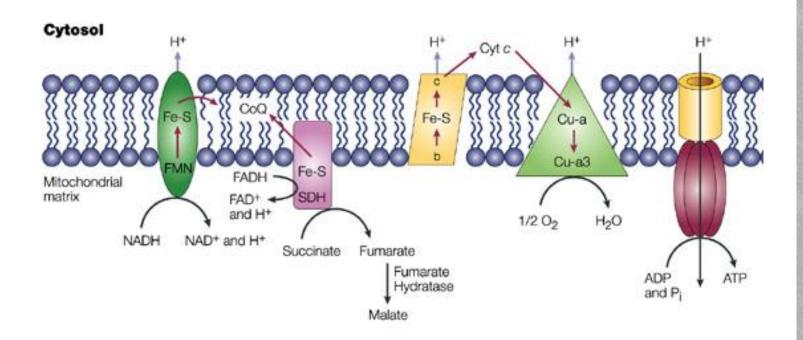






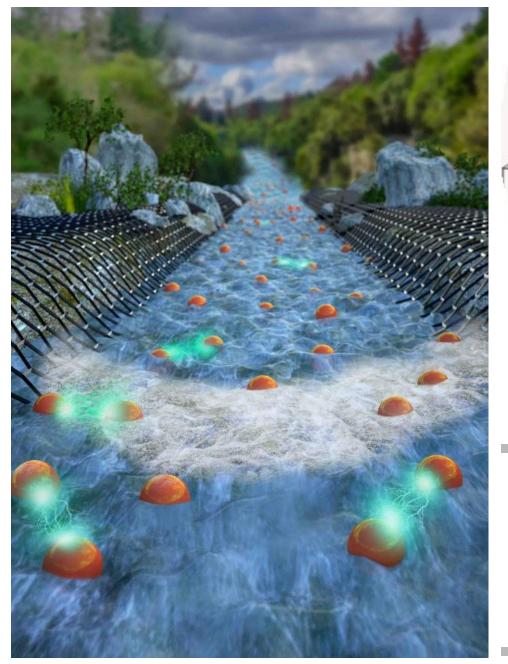
THE OXIDATIVE PHOSPHORYLATION, WHERE ARE WE?

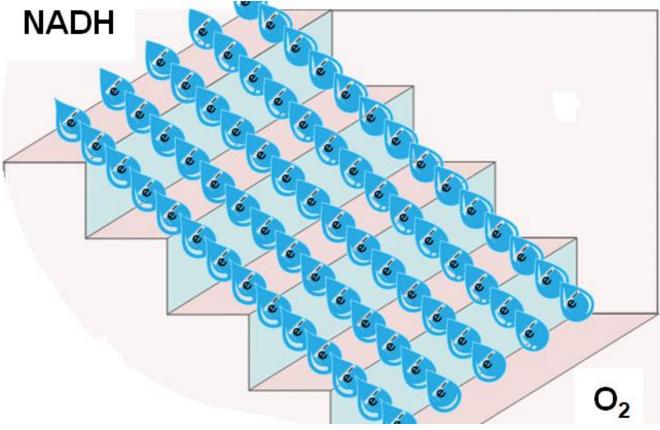
Stages: Digestion; Acetyl-CoA; TCA; OxPhos



(OXPHOS) OXIDATIVE PHOSPHORYLATION

- Generation of ATP aided by O₂ reduction
- Peter Mitchell (1961): the chemiosmotic theory
- 3 major aspects:
- √ (1) Flow of electrons
- (2) The free energy available (exergonic) is <u>coupled to</u> <u>transport of protons</u>
- ✓ (3) Flow of protons provides the free energy for synthesis of ATP (ATP synthase)

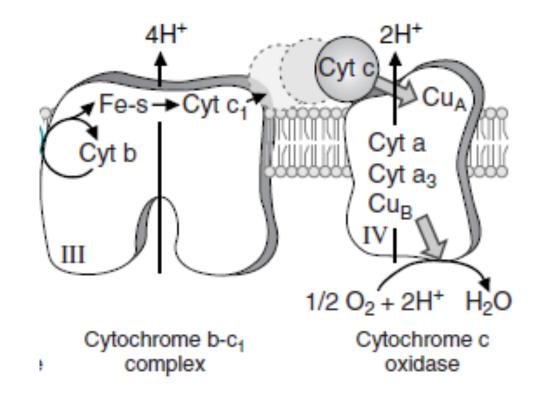




THE ELECTRON TRANSPORT CHAIN: THE RIVER OF ELECTRONS THAT PROVIDES A STEPWISE ENERGY RELEASE

THE CAST OF CHARACTERS: ELECTRON CARRIERS

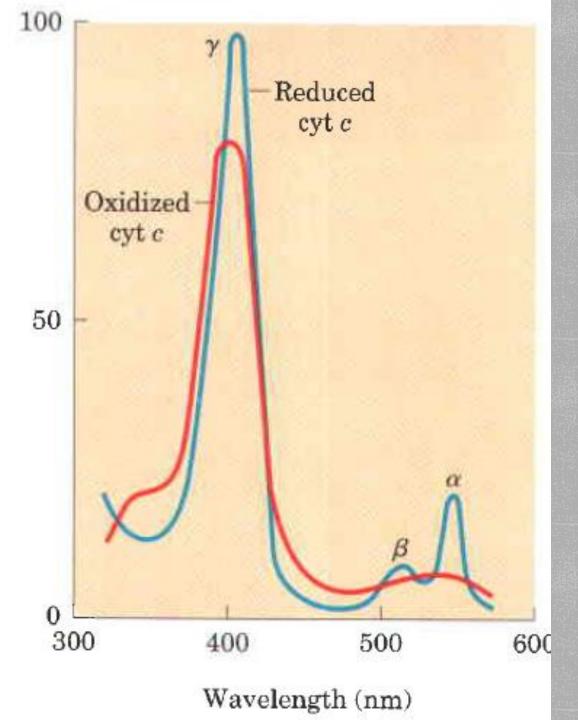




OTHER ELECTRON-CARRYING MOLECULES

"CYTOCHROMES"

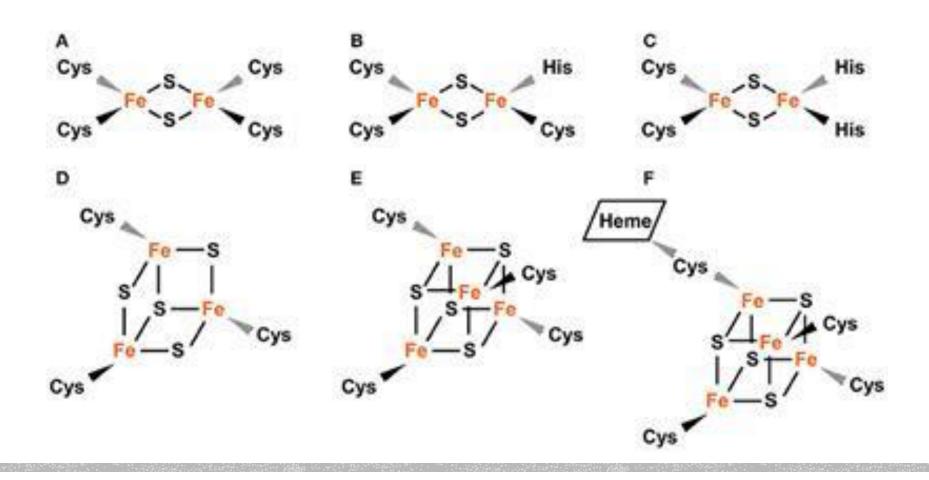
Mitochondria contain three classes of cytochromes (a, b, & c)



OTHER ELECTRON-CARRYING MOLECULES: "CYTOCHROMES"

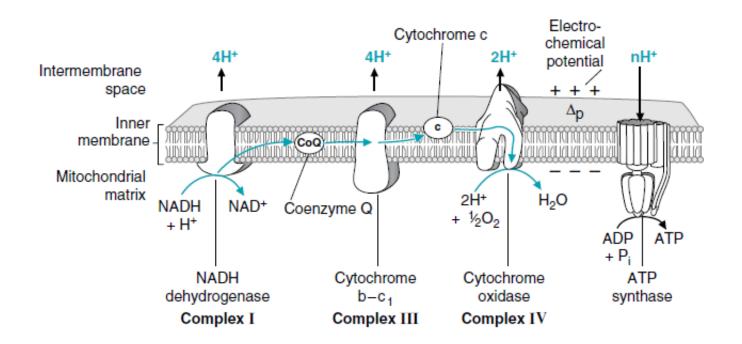
- Light absorption: α band (~600 nm in type a; ~560 nm in type b; ~550 nm in type c)
- Some cytochromes are named by the exact α band wavelength:
 - Cytochrome b₅₆₂; Cytochrome c₅₅₀; Cytochrome c₅₅₁
- Heme can carry one electron
- ΔE^{o} depends on the protein





OTHER ELECTRON-CARRYING MOLECULES: IRON-SULFUR CLUSTERS

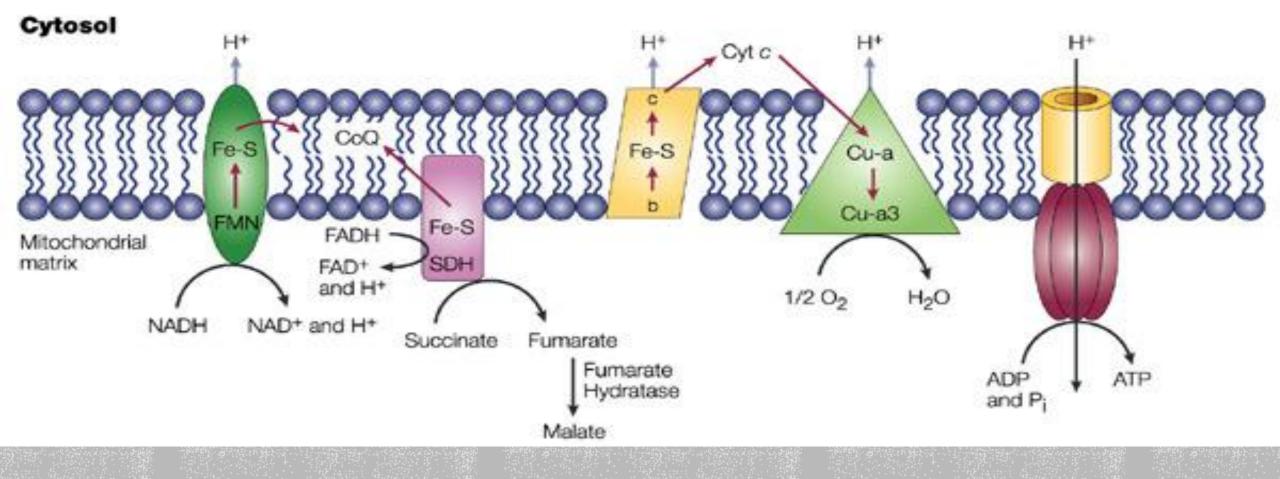




REQUIREMENTS OF OXPHOS

- Electrons' donor (NADH or FADH2) & acceptor (O₂)
- ETC of proteins
- ATP synthase
- An intact IMM





OXIDATIVE PHOSPHORYLATION (OXPHOS)

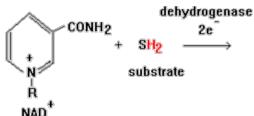


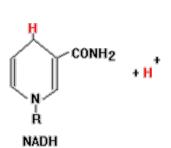
OXI—RED COMPONENTS OF THE ETC: THE NADH DEHYDROGENASE GIANT - COMPLEX I

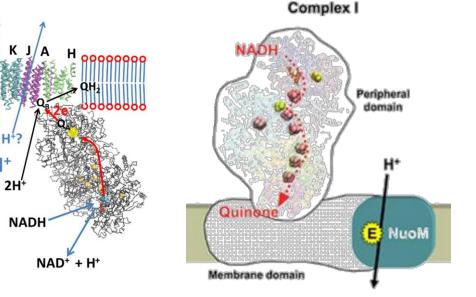
Pr-Cys-S-Fe-S-Cys-Cys-Cys-Pr

- NADH-Q oxidoreductase
- More than 25 polypeptide chain
- A huge flavoprotein membranespanning complex
- FMN is tightly bound
- 7-9 Fe-S centers of at least two different types
- Drop in energy ≈ 13 to 14 kcal
- Binds NADH & CoQ NAD⁺-dependent

4 H⁺



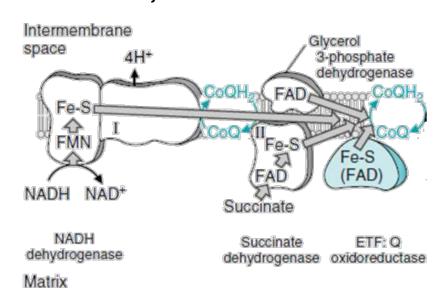


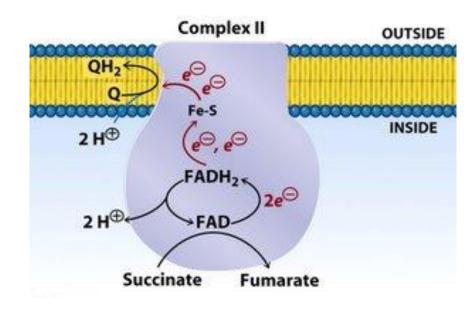




OXI—RED COMPONENTS OF THE ETC: "SUCCINATE DEHYDROGENASE" — COMPLEX II; THE CITRIC ACID CYCLE LINK

- Succinate Dehydrogenase & other flavoproteins
- TCA cycle
 - ✓ETF-CoQ oxidoreductase (ex. fatty acid oxidation); Fatty acyl CoA dehydrogenase; Mitochondrial glycerol phosphate dehydrogenase
 - \approx 0 Kcal, H⁺?









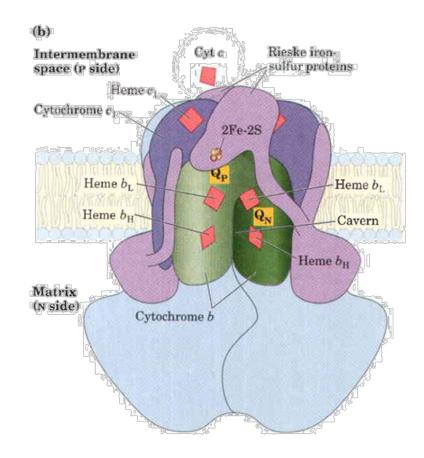


- Lipid-soluble benzoquinone with a long isoprenoid side chain
- Small & hydrophobic (freely diffusible)
- Act at the junction between a 2(e⁻) donor and a 1 (e⁻) acceptor
- Sometimes prescribed for recovering MI patients

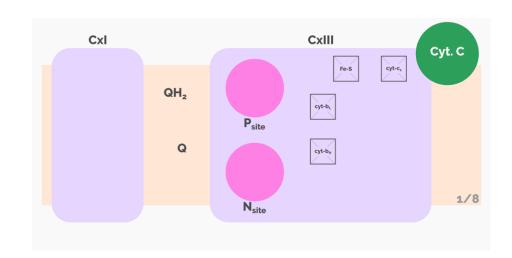
$$\begin{array}{c} \text{CH}_3\text{O} \\ \text{CH}_3\text{O} \\ \text{CH}_2\text{-CH=C-CH}_2)_{10}\text{-H} \\ \text{Ubiquinone (Q)} \\ \text{(fully oxidized)} \\ \text{CH}_3\text{O} \\ \text{CH}_3\text{O} \\ \text{CH}_3\text{O} \\ \text{CH}_3 \\ \text{CH}_3\text{O} \\ \text{CH}_3 \\ \text{OH} \\ \text{CH}_3\text{O} \\ \text{CH}_3 \\ \text{CH}_3$$

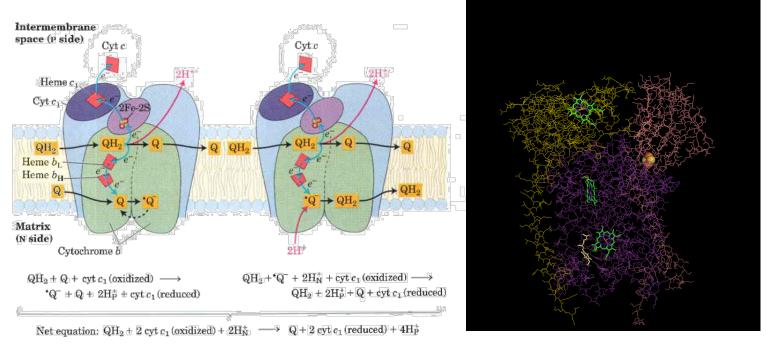
OXI—RED COMPONENTS OF THE ETC: "CYTOCHROME BC₁" — COMPLEX III

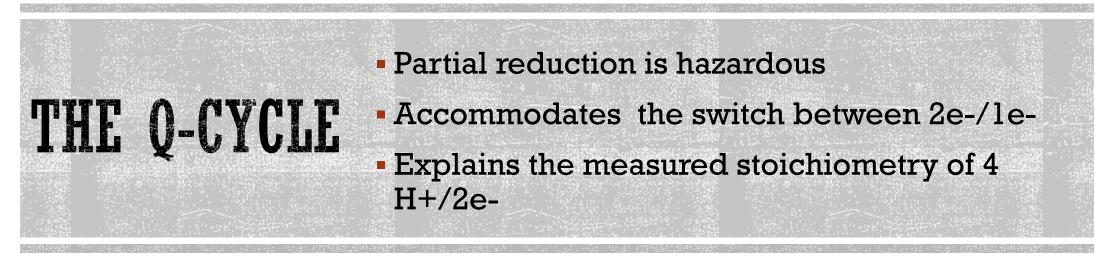
- Q-cytochrome c Oxidoreductase
- 11 subunits including two cytochrome subunits
- Contains iron sulfur center
- Contain three heme groups in two cytochrome subunits
- b_L and b_H in cytochrome b; c type in cytochrome c_I
- Contain two CoQ binding sites
- 4H+



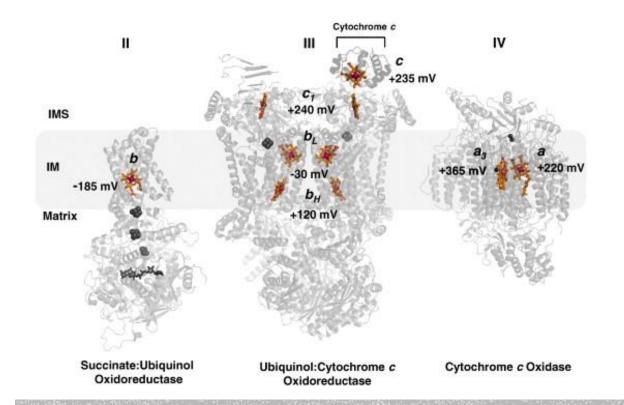


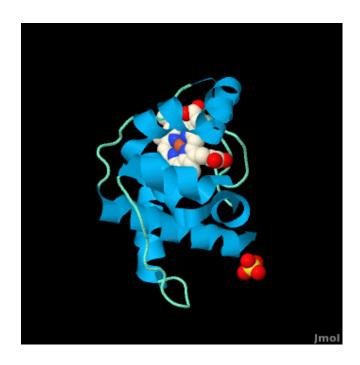










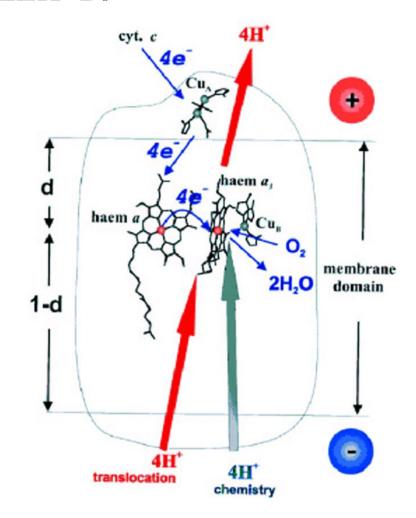


CYTOCHROME C: THE SURFACE MESSENGER



OXI—RED COMPONENTS OF THE ETC: "CYTOCHROME C OXIDASE" — COMPLEX IV

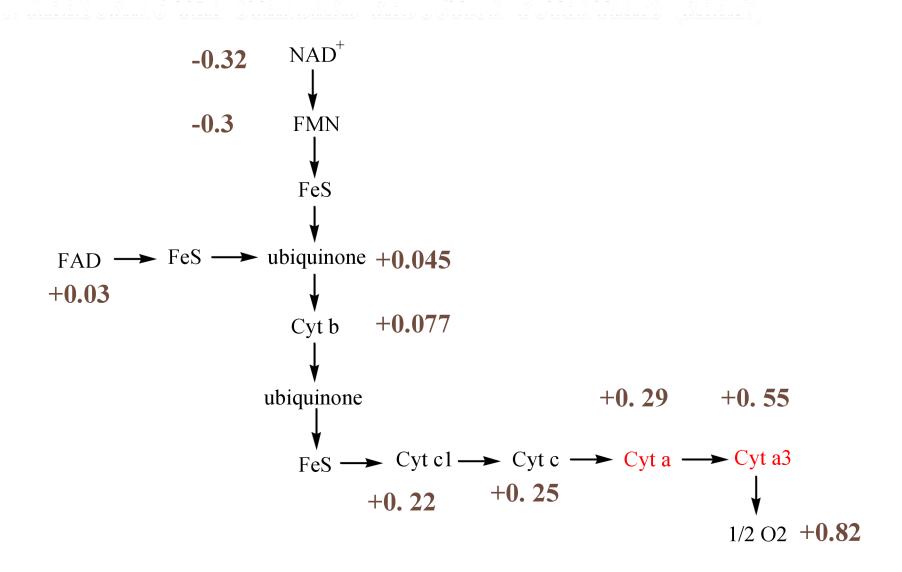
- Contains cytochrome a and a₃
- Contains two copper sites
- Contains oxygen binding sites
- O₂ must accept 4 electrons to be reduced to 2 H₂O (2H⁺/2e-)
- Cytochrome c is one electron carrier
 - Cyt $c_{red} + 4H^+ + O_2 \rightarrow Cyt c_{ox} + 2H_2O$
- Cytochrome oxidase has a much lower
 Km for O₂ than myoglobin (hemoglobin, myoglobin, complex IV)
- Partial reduction of O₂ is hazardous





THE RIGHT ARRANGEMENT: HOW CAN WE PROVE IT?

1. MEASURING THE STANDARD REDUCTION POTENTIALS (REAL?)



THE RIGHT ARRANGEMENT: HOW CAN WE PROVE IT?

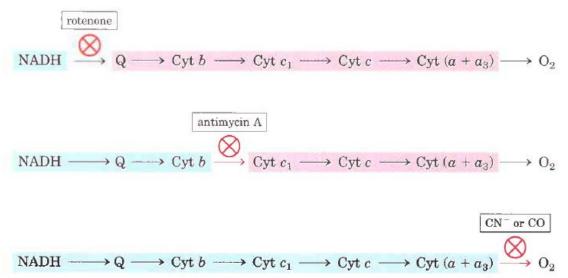
1. MEASURING THE STANDARD REDUCTION POTENTIALS (REAL?)

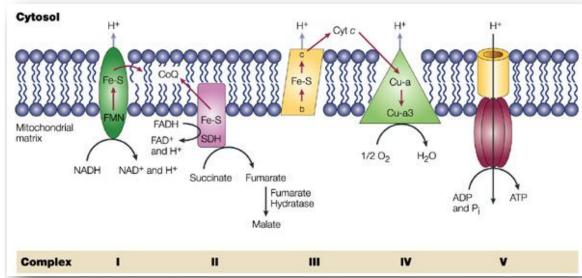
Redox reaction (half-reaction)	$E^{\prime\circ}(V)$
$2H^+ + 2e^- \longrightarrow H_2$	-0.414
$NAD^{+} + H^{+} + 2e^{-} \longrightarrow NADH$	-0.320
$NADP^+ + H^+ + 2e^- \longrightarrow NADPH$	-0.324
NADH dehydrogenase (FMN) + 2H^+ + $2e^ \longrightarrow$ NADH dehydrogenase (FMNH ₂)	-0.30
Ubiquinone + $2H^+ + 2e^- \longrightarrow ubiquinol$	0.045
Cytochrome b (Fe ³⁺) + $e^- \longrightarrow$ cytochrome b (Fe ²⁺)	0.077
Cytochrome $c_{\rm t}$ (Fe ³⁺) + $e^- \longrightarrow$ cytochrome $c_{\rm t}$ (Fe ²⁺)	0.22
Cytochrome c (Fe ³⁺) + $e^- \longrightarrow$ cytochrome c (Fe ²⁺)	0.254
Cytochrome a (Fe ³⁺) + $e^- \longrightarrow$ cytochrome a (Fe ²⁺)	0.29
Cytochrome a_3 (Fe ³⁺) + $e^- \longrightarrow$ cytochrome a_3 (Fe ²⁺)	0.35
$\frac{1}{2}O_2 + 2H^+ + 2e^- \longrightarrow H_2O$	0.8166

THE RIGHT ARRANGEMENT: HOW CAN WE PROVE IT?

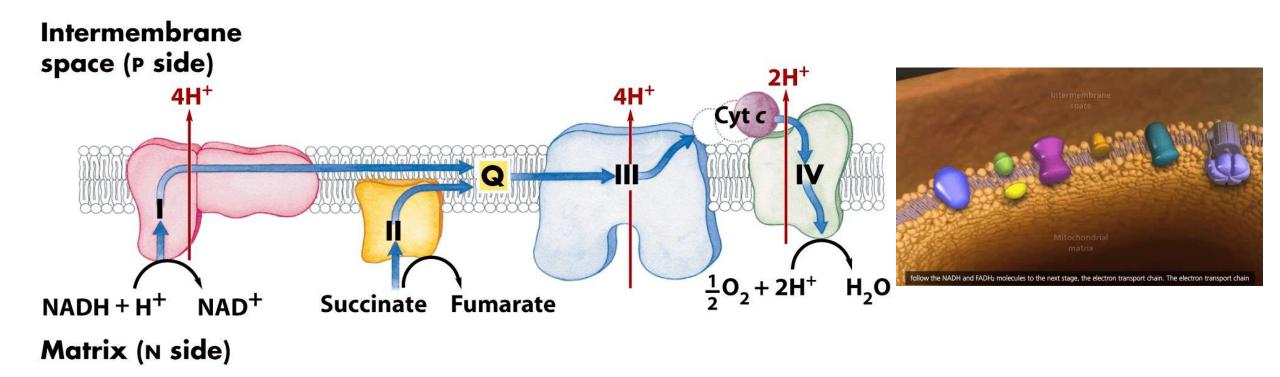
NADH \rightarrow Q \rightarrow cytochrome b \rightarrow cytochrome c1 \rightarrow cytochrome c \rightarrow cytochrome a \rightarrow cytochrome a3 \rightarrow O2

- 2. Reduction of the entire ETC with no O₂
- 3. Addition of inhibitors









PUMPING OF PROTONS

- For every 2 electrons passing:
- 4H⁺ (complex I); 0H⁺ (complex II);
 4H⁺ (complex III), 2H⁺ (complex IV)

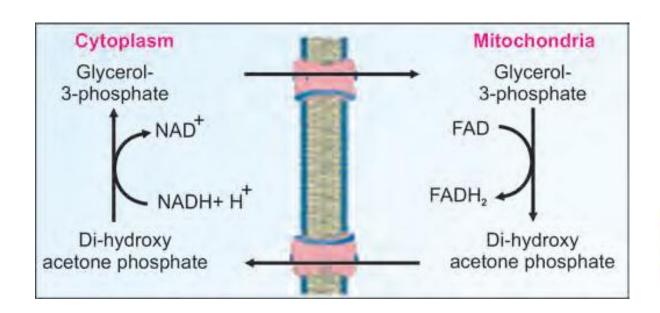


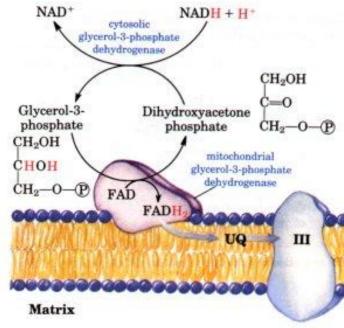
Box 37.3: NAD+ dependent enzymes

- Lactate dehydrogenase (lactate → pyruvate) (see Fig. 9.14)
- Glyceraldehyde-3-phosphate dehydrogenase (glyceraldehyde-3-phosphate → 1,3-bisphosphoglycerate) (see Fig.9.10)
- Pyruvate dehydrogenase (pyruvate → acetyl CoA) (see Fig.9.22)
- Alpha ketoglutarate dehydrogenase (alpha ketoglutarate → succinyl CoA) (see Fig.19.2)
- Beta hydroxyacyl CoA dehydrogenase (beta hydroxyacyl CoA
 → beta ketoacyl CoA (see Step 3, Fig.12.9)
- Glutamate dehydrogenase (Glutamate → alpha ketoglutarate (see Fig.15.9)

SOME NADH PRODUCING ENZYMES







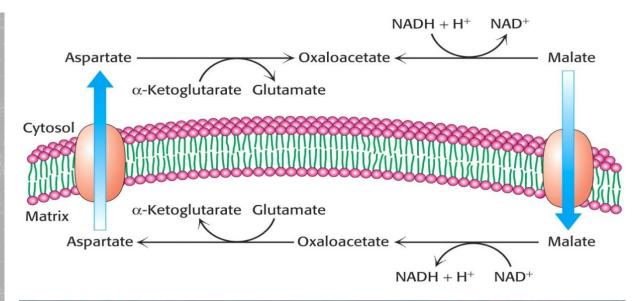
MITOCHONDRIAL SHUTTLING SYSTEMS "CYTOSOLIC NADH"

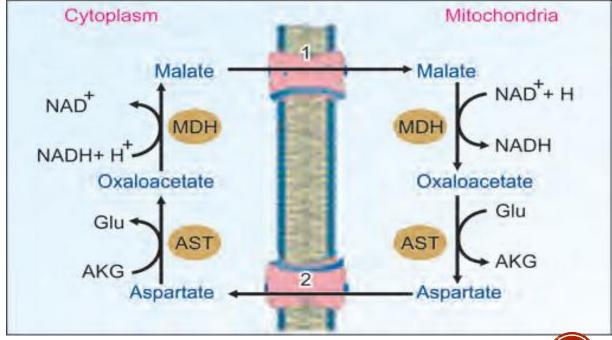
- Glycerol 3-phosphate shuttle
- In skeletal muscle and brain
- Glycolytic pathway as an example
- How NADH passes?
- ATP yield?



MITOCHONDRIAL SHUTTLING SYSTEMS - "CYTOSOLIC NADH"

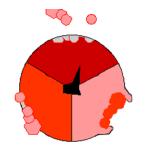
- Malate-Aspartate shuttle
- operates mainly in liver, kidney
- 2 membrane carriers & 4 enzymes
- Readily reversible (vs. Glycerol 3phosphate shuttle)
- NADH can be transferred only if the NADH/NAD+ ratio is higher in the cytosol than in the mitochondrial matrix
- Exchange of key intermediates between mitochondria & cytosol

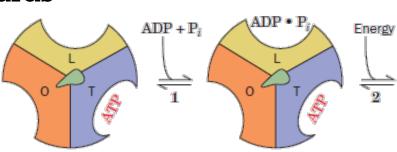


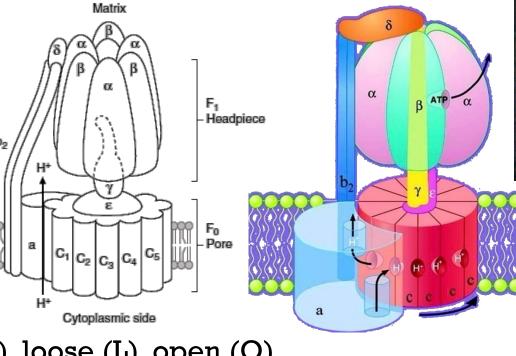


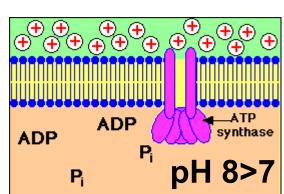
ATP SYNTHASE

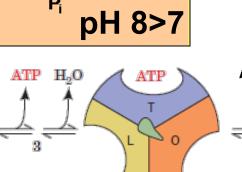
- F1:
 - "γ" subunit: rotates
 - "β" subunit: binds
 - "α" subunit: structural
 - 3 conformations: tight (T), loose (L), open (O)
- F0:
 - "a" subunit: point of entry & exit
 - "c" subunit rotates
 - 4H+/ATP
- Can run backwards

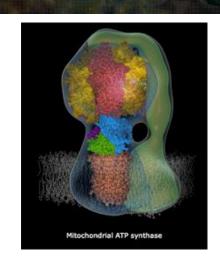


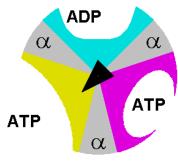




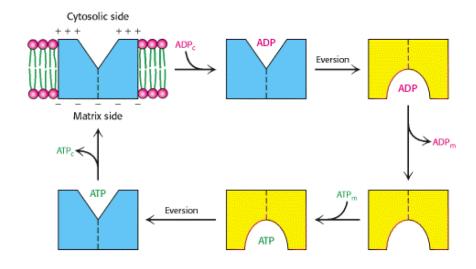


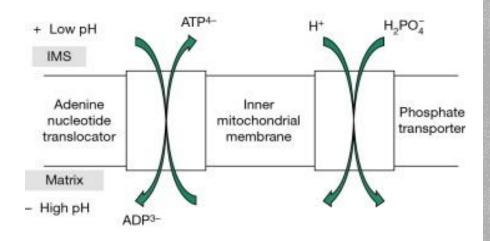








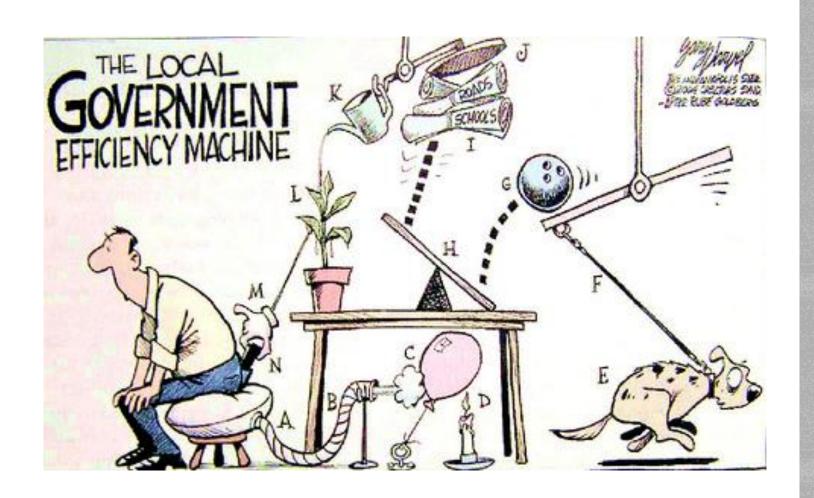




MITOCHONDRIAL SHUTTLING SYSTEMS "ATP/ADP"

- ATP-ADP Translocase (also called adenine nucleotide translocase or ANT)
- The flows of ATP and ADP are coupled (ADP enters only if ATP exits, and vice versa)
- Highly abundant (14% of IMM proteins)
- Contains a single nucleotide-binding site (alternates)
- Similar affinity to ATP and ADP
- Endergonic (25% of ETC)
- Inhibition leads to subsequent inhibition of cellular respiration

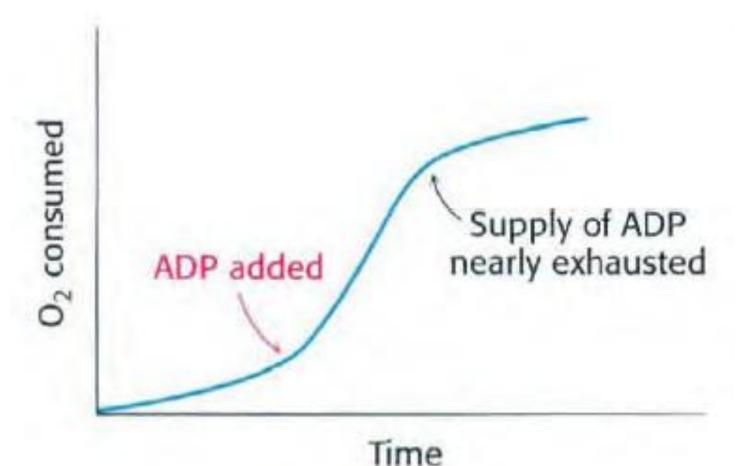




ENERGY YIELD FROM THE ETC

- NADH, -53 kcal, ATP?
- FADH2, -41 kcal, ATP?
- ΔG° is so negative, never reversible
- ATP machine efficiency, (anions, Ca⁺², heat, phosphate, substrates)

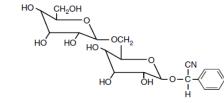




REGULATION — THE NEED FOR ATP

- In skeletal muscles, 20% drop in ATP concentration
- In the heart, Ca⁺² activates
 TCA enzymes for extra push (NADH, ATP), no drop
- ET is tightly coupled to phosphorylation (simultaneously)
- ADP is the most important factor in determining the rate (respiratory control)





Amygdalin, a cyanoglycoside

Anit-cancerous drug

الصقحة الرئيسية > محليات

حراسا نبوز -

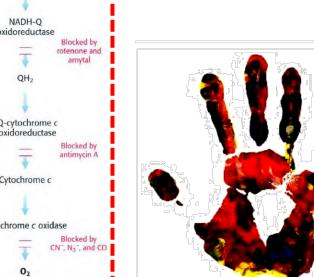
جراسا -نعرض فيما يلي قائمة بأشهر جرائم القتل العائلية التي حدثت في الاردن خلال السنوات الماضية ، والتي كان لكل منها وقع الصدمة حين وقوعها لما تمثله من فعل غريب

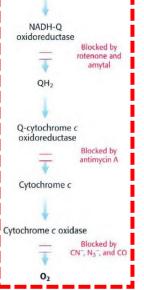
بوضع مادة السيانيد في كأس الحليب وطلب من طفليه ان يشربا منه ، حيث فارقا الحياة بعد 10 دقائق من مغادرة الام المنزل لتعود وتجدهما جثتين هامدنين.

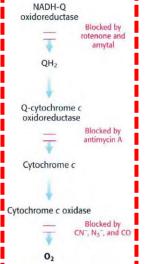
أشهر جرائم القتل العائلية في المملكة

على المجتمع وأعرافه ، فضلا عن مخالفتها الشرائع السماوية والقوانين النافذة والطبيعة الإنسانية بعامة. قضبة السبانيد أول جريمة من نوعها يرتكبها أب ضد ولديه ، اذ قام الاب

وقد ادين الاب بعقوبة الاعدام شنقا الا ان والده اسقط الحق الشخصى كونه وليا عن الطفلين وحكم عليه بالاشغال







Target NADH-Q oxidoreductase

REGULATION — INHIBITION (COUPLING)

Q-cytochrome c oxidoreductase

Cytochrome c oxidase

ATP synthase

Cyanide (CN-), Azide (N3-), & (CO)

Specific inhibitors:

such as amygdalin

√ Cyanoglycosides

are present in

Specific inhibitor

Rotenone (insecticide) &

Amytal (sedative)

Antimycin A (antibiotic)

edible plant pits

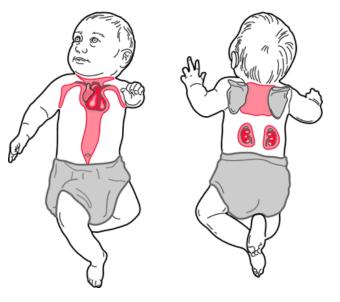
Oligomycin (antibiotic)

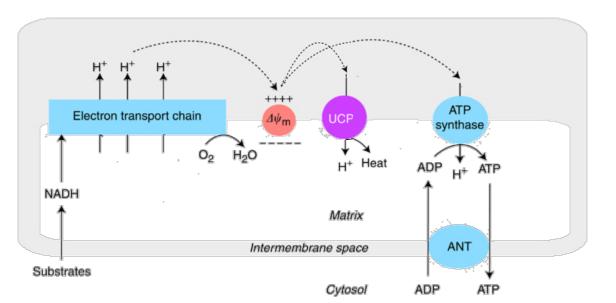




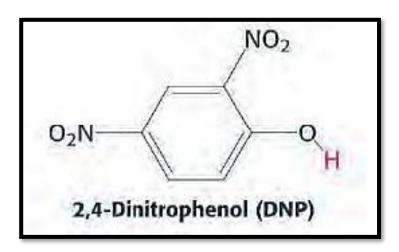
REGULATION — UNCOUPLING PROTEINS (UCPS)

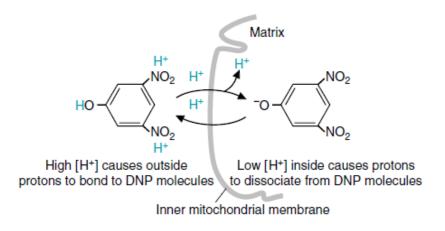
- ➤ Short-circuiting ATP synthase
- ➤ UCP1 (thermogenin):
 - ✓ Brown adipose tissue, nonshivering thermogenesis
 - ✓Infants: neck, breast, around kidneys
 - ✓ Fatty acids directly activates UCP1
- >UCP2 (most cells); UCP3 (skeletal muscle); {UCP4, UCP5} (brain)
- Obesity tendency in some populations





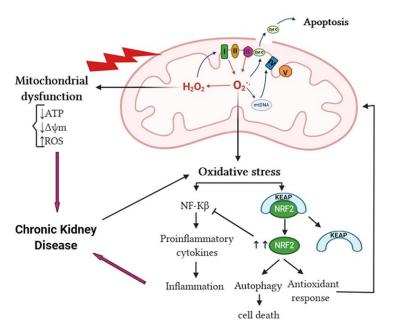


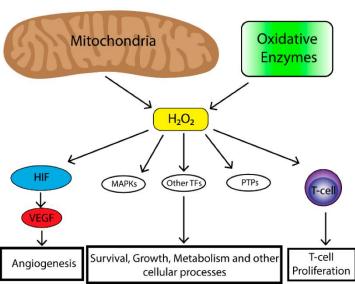




REGULATION — UNCOUPLING UNREGULATED — CHEMICAL UNCOUPLERS

- What is uncoupling?
- How does it occur? Dissipation of PMF
- What is the result?
- Is it physiological or not?
- 2,4-dinitrophenol (DNP) & other acidic aromatic compounds
- What changes happen? ↑ O2 consumption,
 ↑NADH oxidation
- Soviet soldiers were given DNP, FDA banned DNP (1938)

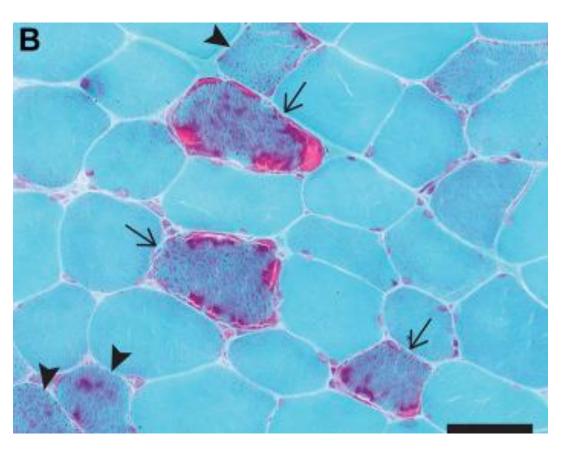




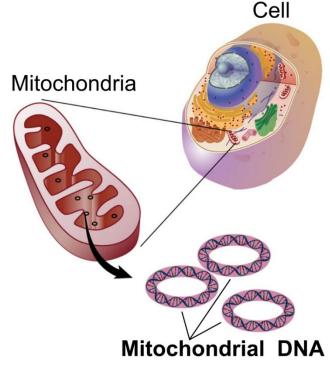
THE DARK SIDE: REACTIVE OXYGEN SPECIES (ROS)

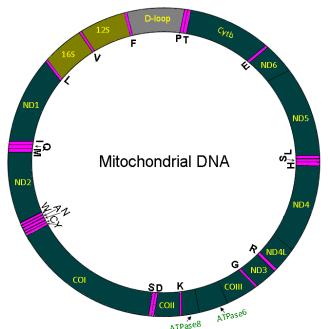
- E.g., FMN, semiquinone in Q cycle
- Single electrons can leak and reduce O_2 to superoxide $(O_2$.
- Pathophysiology:
 - Primary ROS: Superoxide (O₂⁻). Converted by Superoxide Dismutase (SOD) to H₂O₂
 - Damage: lipids (peroxidation), proteins (carbonylation), and DNA (strand breaks)
 - Role in Disease: Contributes to aging, neurodegenerative diseases (Parkinson's, Alzheimer's), ischemia-reperfusion injury, and inflammatory damage

MITOCHONDRIAL DISEASES: GENETICS AND PRINCIPLES



- A muscle biopsy with "ragged red fibers" (Gomori trichrome stain)
- Etiology:
- nDNA or mtDNA
- mtDNA Inheritance: Maternal. High mutation rate, no histones, poor repair.
- Heteroplasmy: A mixture of wild-type and mutant mtDNA within a cell.
- Symptoms appear when the mutant load exceeds a threshold.
- Tissues Affected: High-energy demand tissues: CNS, muscle, heart, liver, kidney.





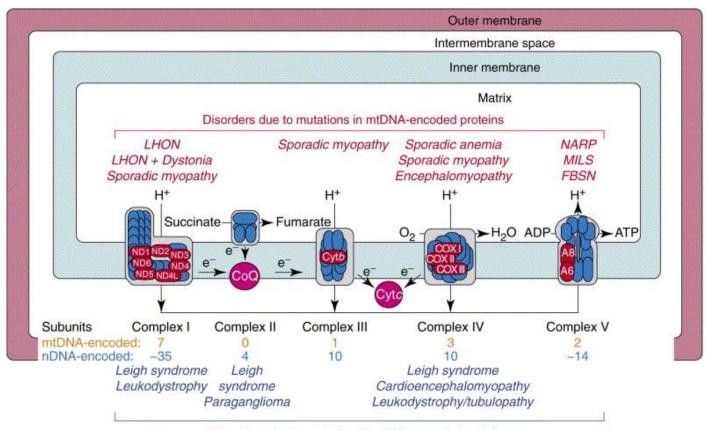
OXPHOS DISEASES — 1 (GENETIC)

- Mitochondrial DNA and OXPHOS Diseases
 - Small (16,569) base pair, double-stranded, circular DNA
 - Encodes 13 subunits: 7 (I), 1 (III), 3 (IV), 2 (F0)
 - Also encodes necessary components for translation of its own mRNA: a large and small rRNA and tRNAs
 - Maternal inheritance, replicative segregation and heteroplasmy
 - Accumulation of somatic mutations with age
 - Highest ATP demands: CNS, heart, skeletal muscle, and kidney, liver



OXPHOS DISEASES - 2 (GENETIC)

- Nuclear Genetic Disorders of Oxidative Phosphorylation
 - 1,000 proteins
 - Usually autosomal recessive
 - Expressed in all tissues
 - Phenotypic expression with high ATP demand



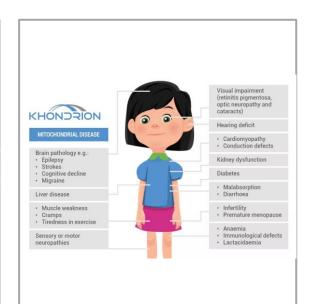
Disorders due to mutations in nDNA-encoded proteins

OXPHOS DISEASES (GENETIC)

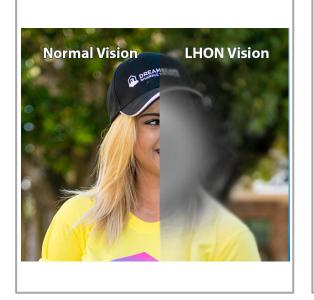


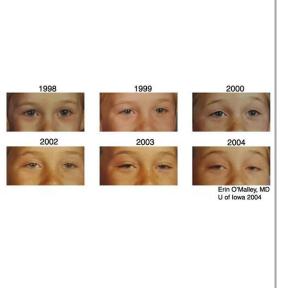
MITOCHONDRIAL DISEASES 2: CLINICAL EXAMPLES

- Examples:
- Leber's Hereditary Optic Neuropathy (LHON): mutation in Complex I. Causes sudden, painless bilateral vision loss in young adults.
- MELAS: Mitochondrial Encephalomyopathy, Lactic Acidosis, and Stroke-like episodes. Caused by mtDNA point mutation.
- Chronic Progressive External
 Ophthalmoplegia (CPEO): ptosis and inability to move the eyes. Often associated with large-scale mtDNA deletions.

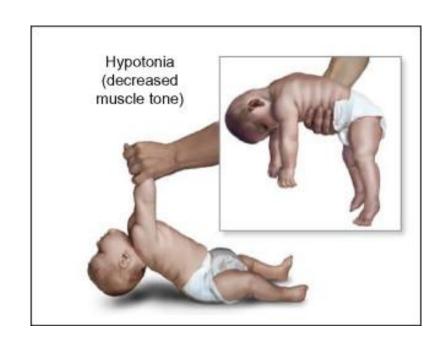












INTERACTIVE CASE STUDY CLINICAL CORRELATION: A DIAGNOSTIC PUZZLE

Presentation:

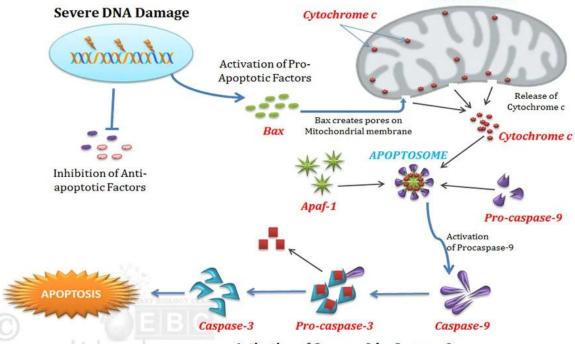
- A 4-year-old boy presents with progressive muscle weakness, exercise intolerance, and developmental delay.
- Lab tests: profound lactic acidosis.
- Muscle biopsy: ragged red fibers.

• Questions:

- What is the most likely category of disease?
 - (Mitochondrial Myopathy)
- What is the biochemical basis for lactic acidosis?
 - (Impaired OXPHOS → anaerobic glycolysis)
- What molecular tests could be ordered?
 - (mtDNA sequencing, nuclear gene panel for OXPHOS proteins)

INTRINSIC PATHWAY OF APOPTOSIS

(Mitochondria Mediated Programmed Cell Death Pathway)



Activation of Caspase-3 by Caspase-9

Mitochondria

BEYOND ENERGY: MITOCHONDRIA IN APOPTOSIS (CELL DEATH) AND AGING

- Apoptosis: In response to severe stress, mitochondria release Cytochrome c and other proteins, which activate caspases and trigger programmed cell death.
- The Mitochondrial Theory of Aging: Accumulation of mtDNA mutations over a lifetime → progressive decline in OXPHOS capacity → increased ROS production → further damage → tissue dysfunction and aging.